

## A TRITERPENE AND SAPONIN FROM ROOTS OF *ILEX PUBESCENS*

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**Key Word Index**—*Ilex pubescens*; Aquifoliaceae; triterpene saponin; pentacyclic triterpene; ilexaponin A1; ilexgenin A.

**Abstract**—A new triterpene and its glycoside have been isolated from methanol extract of roots of *Ilex pubescens*. Their structures were established as 3 $\beta$ , 19 $\alpha$ -dihydroxyurs-12-en-24,28-dioic acid and its 28- $\beta$ -D-glucopyranosyl ester, based on chemical and spectral evidence.

### INTRODUCTION

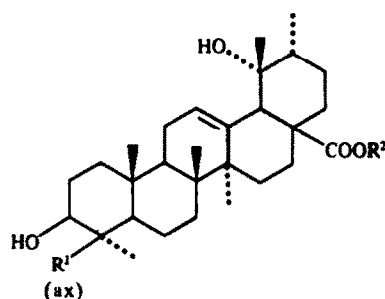
In China, 'mao-dong-qing' the root of *Ilex pubescens* Hook. et Arn. is widely used for the treatment of cardiovascular diseases and hypercholestaemia. Some phenolic compounds have been isolated from this plant [1]. We have investigated the constituents of the roots of this plant, and isolated a new triterpene and its glycoside. We have named them ilexgenin A and ilexaponin A1, and established their structures as 1 and 2, respectively.

### RESULTS AND DISCUSSION

The dried roots of *Ilex pubescens* were extracted with benzene, followed by methanol. The methanol extract was suspended in water, then extracted successively with benzene and ethyl acetate. From the ethyl acetate fraction, two compounds, 1 and 2, were isolated by reversed phase and normal phase column chromatography.

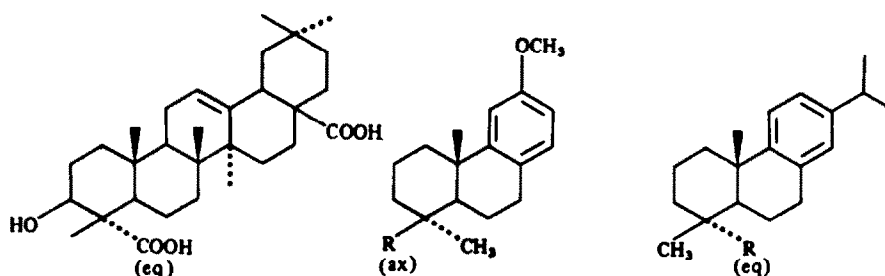
Ilexgenin A (1), C<sub>30</sub>H<sub>46</sub>O<sub>6</sub>, had mp > 300°. Its <sup>13</sup>C NMR spectrum revealed thirty carbon signals (CH<sub>3</sub>- × 6, -CH<sub>2</sub>- × 9, >CH- × 4, >C< × 5, >CH-O × 1, >C=O × 1, >C=CH- × 1, CO × 2). Its IR spectrum showed bands at 3450 cm<sup>-1</sup> for a hydroxyl group, at 1690 and 1685 cm<sup>-1</sup> for a carboxyl group and at 1630 cm<sup>-1</sup> for

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	R <sup>1</sup>	R <sup>2</sup>
1	COOH	H
1a	COOCH <sub>3</sub>	CH <sub>3</sub>
1b	COOCH <sub>3</sub>	H
2	COOH	Glc
2a	COOCH <sub>3</sub>	Glc
3	CH <sub>3</sub>	H
3a	CH <sub>3</sub>	CH <sub>3</sub>

Glc:  $\beta$ -D-glucopyranosyl



4

5 R = CH<sub>3</sub>  
6 R = COOCH<sub>3</sub>

7 R = CH<sub>3</sub>  
8 R = COOCH<sub>3</sub>

a double bond. The EI mass spectrum showed a peak at  $m/z$  264, the characteristic retro-Diels–Alder cleavage peak of an olean-12-en or urs-12-en-28-oic acid derivative which possess a hydroxy group on either the D or E ring. The  $^1\text{H}$  NMR spectrum showed the peak at  $\delta$  3.08 (1H, s, H-18), which suggested the presence of the 19-*O*-substituted urs-12-ene skeleton [2]. The  $^{13}\text{C}$  NMR spectrum of the methyl ester (1a) was compared with methyl pomolate (3a) (Table 1). The spectrum of 1a showed one less methyl and one more carboxyl and methoxyl signals than that of 3a. The rest of the spectrum was very similar to that of 3a. In particular, carbon signals due to the D and E rings of 3a appeared at the corresponding position in the spectrum of 1a, while the signals of the A and B ring carbons were observed at somewhat different positions. In

the EI mass spectrum of 1a, retro-Diels–Alder cleavage peaks from the C and D rings appeared at  $m/z$  278, 260 and 201 as well as in the spectrum of 3a. However, a peak at  $m/z$  208, arising by retro-Diels–Alder cleavage peak from the A and B rings of a pomolic acid (3)-type (non-substituted gem-dimethyl) triterpene, was not observed but a peak at  $m/z$  252 was detected showing that one of the methyl groups at C-4 were replaced by  $\text{COOCH}_3$  (Scheme 1). The  $^1\text{H}$  NMR spectrum of 1a showed the presence of a  $3\beta$ -hydroxy by the signal at  $\delta$  3.14 (1H, dd,  $J = 4, 11$  Hz).

These facts indicate that 1a is related to 3a. The  $^{13}\text{C}$  NMR spectrum of 1a showed that the signal of the tertiary C-4 was shifted significantly downfield by 10.5 ppm, while the other tertiary carbons, C-8 and 10 were not. Therefore, the  $\text{COOCH}_3$  group should be restricted to the C-23 or C-24 positions. The  $^{13}\text{C}$  NMR spectrum of 1a was compared with that of gypsogenic acid (4) [4] which has a C-23 equational–COOH group which results in significant differences in the chemical shifts of the A and B ring carbons. (Table 1). Since no  $^{13}\text{C}$  NMR data for a triterpene with a C-24 carboxy methyl ester group were available, the data of several diterpenes (5–8) which have similar partial structures were compared [5]. Substitution ( $\text{CH}_3 \rightarrow -\text{COOMe}$ ) induced shift values in the  $^{13}\text{C}$  NMR spectra for 5 and 6 corresponded to the values from 3a and 1a, while these for 7 and 8 was quite different (Table 2). These facts prove the location of the carbomethoxyl group in 3 is the C-24 (axial) position. Therefore, the structure of 1 is elucidated to be  $3\beta,19\alpha$ -dihydroxyurs-12-en-24,28-dioic acid.

A new saponin, ilexaponin A1 (2), was obtained as an amorphous white powder. The IR spectrum of 2 exhibited hydroxyl ( $3400\text{ cm}^{-1}$ ), carboxy ( $1685\text{ cm}^{-1}$ ), and ester ( $1720\text{ cm}^{-1}$ ) bands. The  $^{13}\text{C}$  NMR spectrum of 2 showed characteristic signals of an ester-linked  $\beta$ -glucopyranose moiety [6]. The rest of the  $^{13}\text{C}$  NMR signals were essentially the same as those of 1, except one of the carbonyl signals, which was shifted downfield by 3.5 ppm. The aglycone of 2 was obtained by alkaline hydrolysis and identified as 1 by means of  $^{13}\text{C}$  NMR spectroscopy. Acid hydrolysis of 2 afforded glucose, identified by GLC. The  $^1\text{H}$  NMR spectrum of 2 supported the  $\beta$ -anomeric configuration ( $\delta$  6.29, 1H, d,  $J = 6.8$  Hz). To decide the location of the glucosyl ester, 2 was methylated by diazomethane followed by alkaline hydrolysis to afford 1b [7]. The EI mass spectrum of 1b showed fragment peaks at  $m/z$  470, 452, 398, 264, 246, 252 which indicated that the methyl group was located on the C-24 carboxyl. On the contrary, peaks at  $m/z$  456, 438, 384, 278, 260, 238, which would appear if there was a C-28 COOMe group, were not observed (Scheme 2). Therefore, the location of the glucosyl residue of 2 was deduced to be on the C-28 carboxyl group. It follows that 2 can be formulated as  $3\beta,19\alpha$ -dihydroxyurs-12-en-24,28-dioic acid-28- $\beta$ -D-glucopyranosylester.

## EXPERIMENTAL

**General procedure.** NMR spectra were taken at  $25^\circ$  using TMS as internal standard;  $^{13}\text{C}$  NMR at 25.15 MHz and  $^1\text{H}$  NMR at 100 MHz. EIMS were taken at 75 eV. Mps were taken on a micro hot-stage and are uncorr.

**Plant material.** A sample of 'mao-dong-qing', the root of *Ilex pubescens*, was purchased in 1983 from Mikuni Co. Ltd (Osaka).

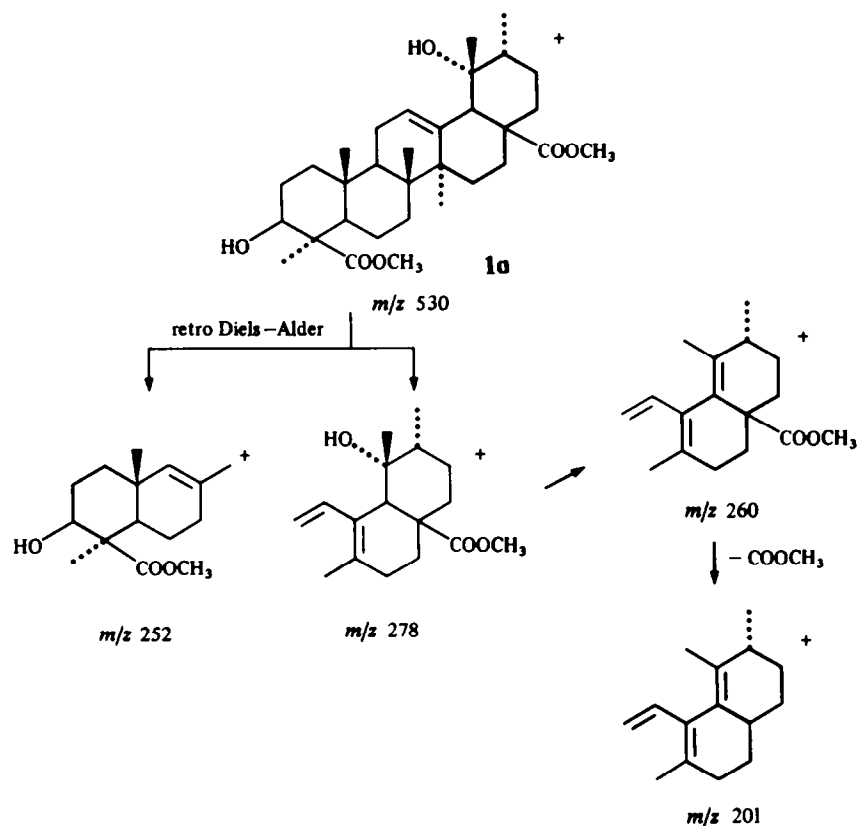
Table 1.  $^{13}\text{C}$  NMR spectral data ( $\delta$ ) for 1, 2 and related compounds

Carbon	3a*	1a	1	2	4†
<i>Genin</i>					
C-1	38.7	39.1	39.8	39.7	37.6
C-2	27.4	28.1	29.1	29.0	26.2
C-3	79.0	78.3	78.3	78.3	74.0
C-4	38.5	49.0	49.2	49.1	53.0
C-5	55.2	56.5	56.9	56.9	50.5
C-6	18.4	20.2	20.9	21.0	20.2
C-7	32.8	33.0	33.9	33.8	31.5
C-8	39.9	39.7	40.2	40.3	38.6
C-9	47.2	46.6	47.2	47.1	46.9
C-10	36.9	37.2	37.9	37.8	35.3
C-11	23.7	23.8	24.5	24.4	22.3
C-12	129.1	129.1	128.1	128.4	122.4
C-13	138.0	138.0	139.9	139.2	143.3
C-14	41.1	41.2	42.2	42.1	40.7
C-15	28.1	28.2	29.1	29.1	26.8
C-16	25.5	25.5	26.5	26.0	22.2
C-17	47.9	47.9	48.3	48.6	45.1
C-18	53.2	53.3	54.7	54.4	40.5
C-19	73.1	73.1	72.7	72.6	45.0
C-20	41.1	41.1	42.2	42.1	29.5
C-21	26.0	26.0	27.0	26.6	32.7
C-22	37.4	37.4	38.4	37.8	31.5
C-23	28.1	23.6	24.2	24.4	179.2
C-24	15.2	178.5	180.6	181.0	10.7
C-25	15.5	13.1	13.9	13.9	14.5
C-26	16.6	16.5	17.1	17.3	15.9
C-27	24.5	24.2	24.5	24.7	24.7
C-28	178.3	178.3	180.6	177.1	178.7
C-29	27.2	27.4	27.0	27.0	31.8
C-30	16.1	16.1	16.8	16.7	22.3
$\text{COOH}_3$	51.5	51.5			
		51.2			
<i>Sugar</i>					
G-1				95.7	
G-2				73.8	
G-3				78.7	
G-4				71.1	
G-5				79.1	
G-6				62.2	

\*Cited from ref. [3].

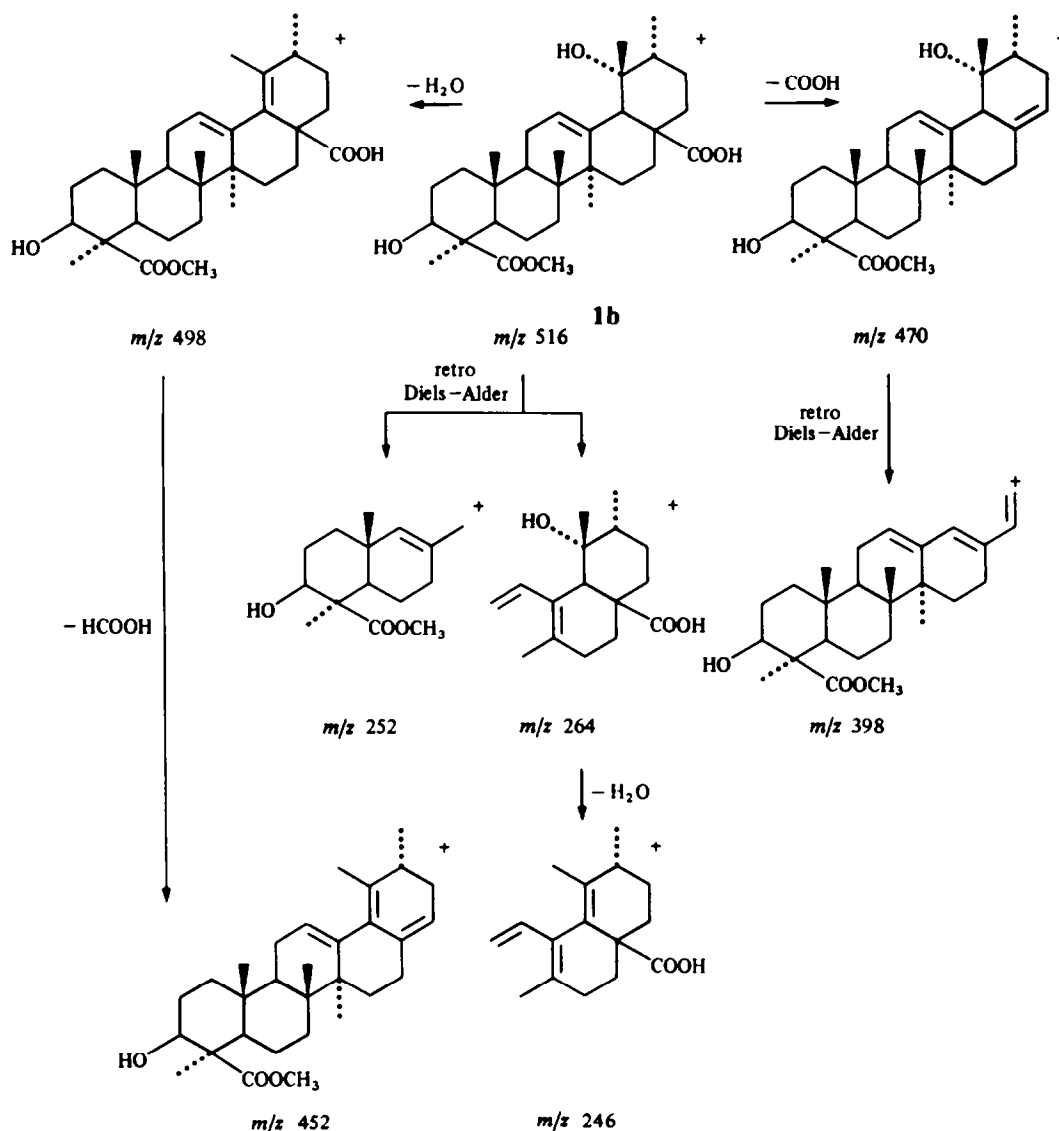
†Cited from ref. [4].

3a, 1a, in  $\text{CDCl}_3$ ; 1, 2, 4, in  $\text{C}_2\text{D}_2\text{N}$ .

Table 2.  $^{13}\text{C}$ NMR substitution induced shift ( $\Delta\delta$ ) from **3a** to **1a**, and related compounds

Carbon	5	6	$\Delta\delta$ 6-5	7	8	$\Delta\delta$ 8-7	$\Delta\delta$ 1a-3a
C-1	38.9	39.4	+0.5	38.9	38.0	-0.9	+0.4
C-2	19.3	20.2	+0.9	19.3	18.6	-0.7	+0.7
C-3	41.7	37.6	-4.1	41.8	36.7	-5.1	-0.7
C-4	33.5	44.0	+10.5	33.4	47.7	+14.3	+10.5
C-5	50.3	52.8	+2.5	50.4	44.9	-5.6	+1.3
C-6	19.2	21.1	+1.9	19.1	21.7	+2.6	+1.8
C-7	29.6	31.2	+1.6	30.5	30.0	-0.5	+0.2
C-8	127.5	127.6	+0.1	134.8	134.7	-0.5	-0.2
C-9	151.4	149.3	-2.1	147.6	146.9	-0.7	-0.6
C-10	38.0	38.6	+0.6	37.5	37.0	-0.5	+0.3
C-11	110.1	111.2		124.2	124.1		
C-12	157.7	157.8		123.8	123.9		
C-13	110.7	111.2		145.3	145.7		
C-14	129.7	129.8		126.8	126.9		
C-15				33.5	33.5		
C-16				24.0	24.0		
C-17				24.0	24.0		
C-18	33.3	28.5	-4.8	33.3	179.1	(C-23)	-4.5
C-19	21.7	177.8		21.6	16.5	-5.1	(C-24)
C-20	24.7	22.9	-1.8	24.9	25.1	+0.2	(C-25)
COOH <sub>3</sub>		51.2			51.9		

 $\delta$  Values of 5-8 were cited from ref. [5].



Scheme 2.

**Extraction of triterpene and glycosides.** Dried roots (4.7 kg) were crushed and extracted with  $C_6H_6$  (20 l.  $\times$  2). The residue was re-extracted with MeOH (20 l.  $\times$  4) and the MeOH extract was taken to dryness (414 g). A suspension of the resulting extract in  $H_2O$  was washed with  $C_6H_6$ , then extracted with EtOAc to give an EtOAc extract (103 g). The EtOAc extract (100 g) was chromatographed on silica gel (Kieselgel 60, Merck, 3 kg) developing with  $CHCl_3$ -MeOH (10:1) and  $CHCl_3$ -MeOH- $H_2O$  (15:2:0.1-15:10:2) mixture successively to yield 17 fractions.

From fraction 3 (eluate of  $CHCl_3$ -MeOH), compound 1 (370 mg) was obtained. A portion (1.6 g) of fraction 6 (26 g, eluate of  $CHCl_3$ -MeOH- $H_2O$ , 15:2:0.1) was subjected to chromatography on silica gel using EtOAc- $CHCl_3$ -EtOH- $H_2O$ , followed by reversed phase chromatography on ODS using MeOH- $H_2O$  to afford compound 2 (585 mg). From a portion (1.6 g) of fraction 4 (11 g) eluted with  $CHCl_3$ -MeOH- $H_2O$  (15:2:0.1), compound 2 (139 mg) was also obtained by a similar procedure to that above.

**Compound 1.** White powder,  $[\alpha]_D^{20} + 30.8^\circ$  (pyridine:  $c$  0.97). (found: C, 71.00; H, 9.33;  $C_{30}H_{46}O_6$  requires: C, 71.68; H, 9.22%);

EIMS  $m/z$  (rel. int.): 502  $[M]^+$  (4), 484  $[M-H_2O]^+$  (34), 456  $[M-HCOOH]^+$  (29), 438  $[M-HCOOH-H_2O]^+$  (15), 264 (8), 238 (21), 146 (100); HRMS  $m/z$ : 502.3295  $[M]^+$  (calc. for  $C_{30}H_{46}O_6$ : 502.3294); IR  $\nu_{max}^{Nujol}$   $cm^{-1}$ : 3450, 1690, 1685, 1630;  $^1H$  NMR (pyridine- $d_5$ ):  $\delta$  1.16 (3H, s, Me), 1.18 (3H, s, Me), 1.48 (3H, s, Me), 1.74 (3H, s, Me), 1.78 (3H, s, Me), 1.14 (3H, d,  $J = 7.8$  Hz, 30 Me), 3.08 (1H, s, H-18), 3.38 (1H, dd,  $J = 4.3$ , 11.6 Hz, H-3), 5.14 (1H, br s, 19-OH), 5.65 (1H, br t, H-12);  $^{13}C$  NMR: Table 1.

**Methylation of compound 1.** Compound 1 (75 mg) was treated with diazomethane followed by chromatography to give 1a (di-Me ester); mp 199-203° (from MeOH);  $[\alpha]_D^{20} + 78.2^\circ$  (pyridine,  $c$  1.01); (found: C, 72.27; H, 9.55;  $C_{32}H_{50}O_6$  requires: C, 72.40; H, 9.50%); HRMS  $m/z$ : 530.3597  $[M]^+$  (calc. for  $C_{32}H_{50}O_6$ : 530.3607); EIMS  $m/z$  530  $[M]^+$ , 512, 470, 278, 252, 260, 201;  $^1H$  NMR ( $CDCl_3$ ):  $\delta$  0.70 (3H, s, Me), 0.76 (3H, s, Me), 1.21 (3H, s, Me), 1.26 (3H, s, Me), 1.40 (3H, s, Me), 0.94 (3H, d,  $J = 6.6$  Hz, 30-Me) 2.60 (1H, s, H-18), 3.14 (1H, dd,  $J = 4$ , 11 Hz, H-3), 3.60 (3H, s, OMe), 3.67 (3H, s, OMe), 5.35 (1H, t,  $J = 3.5$  Hz, H-12);  $^{13}C$  NMR: Table 1.

**Compound 2.** White powder,  $[\alpha]_D^{20} + 25.5^\circ$  (pyridine;  $c$  0.69). (found: C, 62.06; H, 8.43;  $C_{36}H_{56}O_{11} \cdot 2H_2O$  requires: C, 61.69; H, 8.63 %). EIMS (acetate)  $m/z$ : 331,  $[Glc(Ac)_4]^+$ ; IR  $\nu_{max}^{Nujol}$   $cm^{-1}$ : 3400, 1720, 1685;  $^1H$  NMR (pyridine- $d_5$ ):  $\delta$  1.18 (3H, s, Me), 1.29 (3H, s, Me), 1.44 (3H, s, Me), 1.73 (6H, s, Me), 1.09 (3H, d,  $J = 6.1$  Hz), 2.97 (1H, s, H-18), 3.30 (1H, dd,  $J = 4, 11$  Hz, H-3), 5.21 (1H, br s, 19-OH), 5.61 (1H, br t, H-12), 6.29 (1H, d,  $J = 6.8$  Hz, anomeric H). Mineral acid hydrolysis of 2 afforded D-glucose, identified by GC as its trimethylsilyl derivative.

**Alkaline hydrolysis of compound 2.** A soln of 2 (25 mg) in 20% KOH (1 ml) was heated at  $80^\circ$  for 15 min. The soln was neutralized with 5% HCl, diluted with  $H_2O$ , and extracted with  $Et_2O$ . The  $Et_2O$  extract was chromatographed on silica gel using  $CHCl_3$ -MeOH- $H_2O$  (15:1:0.1) to afford a white powder (19 mg), identified as 1 by  $^1H$  NMR and  $^{13}C$  NMR.

**Methylation of compound 2 followed by alkaline hydrolysis for MS.** A MeOH soln of 2 (30 mg) was treated with diazomethane and the resulting methyl ester was hydrolysed in KOH soln, yielding 1b (8 mg).

**Compound 1b.** White powder, HRMS  $m/z$ : 516.3482  $[M]^+$  (calc. for  $C_{31}H_{48}O_6$ : 516.3450); EIMS  $m/z$ : 516  $[M]^+$  498, 470, 452, 398, 264, 252, 246, 234, 218, 201, 187, 175, 146.  $^1H$  NMR (in pyridine- $d_5$ ):  $\delta$  0.87, 1.14, 1.48, 1.59, 1.75 (each 3H, s, H-23, 25, 26, 27, 29) 1.14 (3H, d,  $J = 7.8$  Hz, H-30); 3.10 (1H, s, H-18); 3.30 (1H,

dd,  $J = 4, 11$  Hz, H-3) 3.57 (3H, s, OMe); 5.11 (1H, br s, OH); 5.64 (1H, t,  $J = 3.5$  Hz, H-12).

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